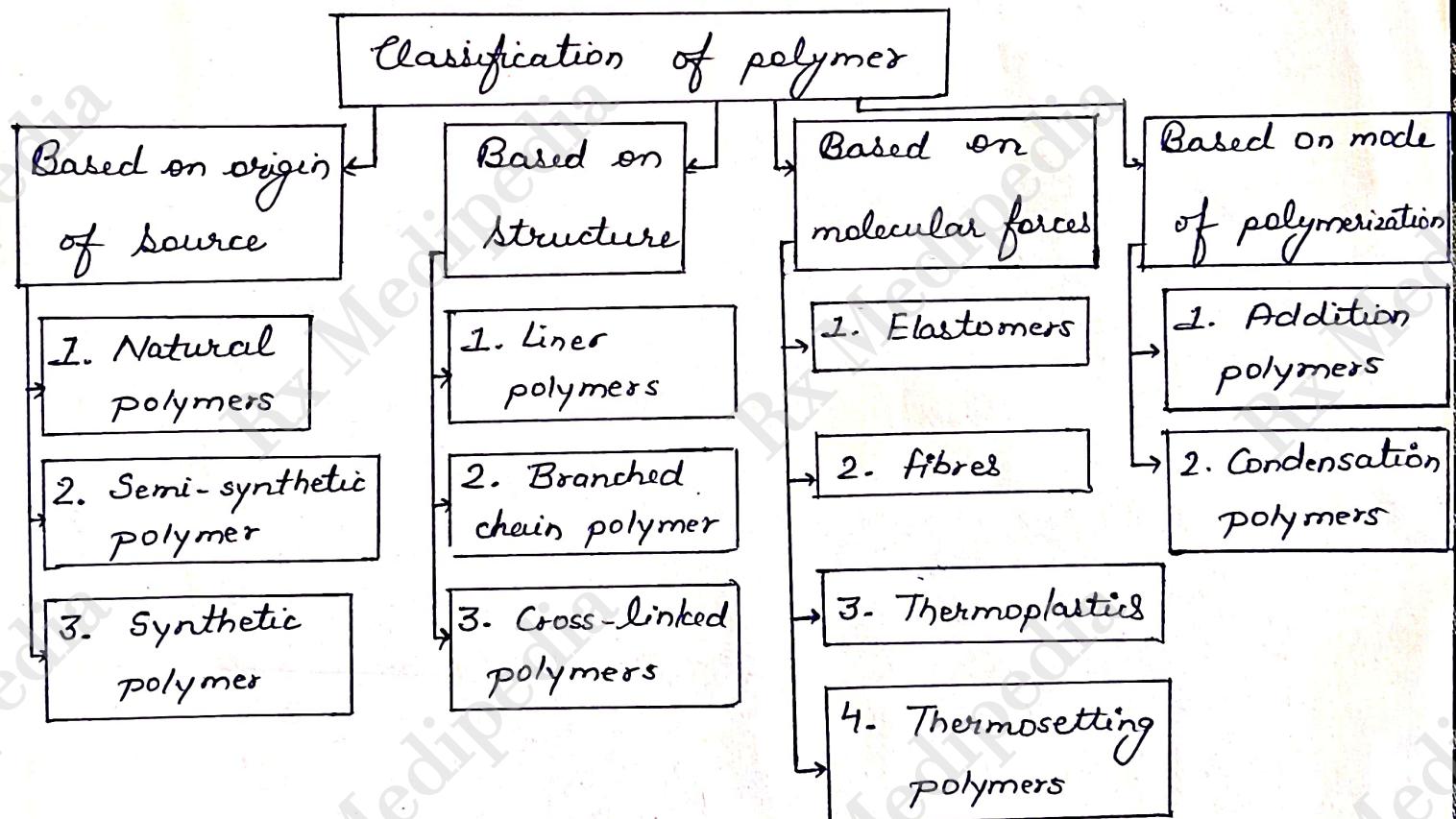


# 'POLYMERS'

## INTRODUCTION

Polymers are the compounds which have high molecular masses and formed by monomers. In greek, the word poly means 'many', and mers means 'units of parts'.

## CLASSIFICATION



## 1. Based on Source

### a) Natural polymer

These are derived from Natural sources and can be polysaccharides and protein in chemical Nature.

ex. Albumin, Starch etc.

### b) Semi-Synthetic polymers

These are derived from naturally occurring polymers by chemical modifications.

ex. Vulcanized rubber, Gun cotton etc.

### c) Synthetic polymers

These are artificial origin which consist of fibres. They are prepared by Laboratory.

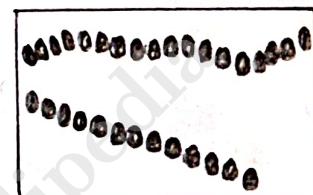
ex. Buna-S, Buna-R etc.

## 2. Based on Structure

### a) Linear polymers

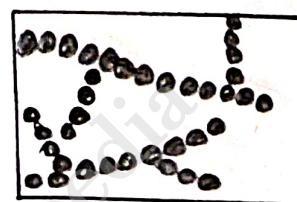
The smallest repeating unit arranged in straight line path is known as linear polymer.

ex. PVC



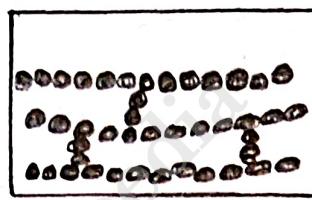
### b) Branched chain polymers

They contain chains having some branches. for ex. low density polymer, Polyethylene.



### c) Cross linked chain polymer

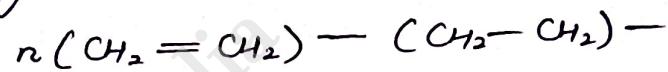
In this, all molecules are chemically bonded together which forms three-dimensional network.



### 3. Based on Polymerization

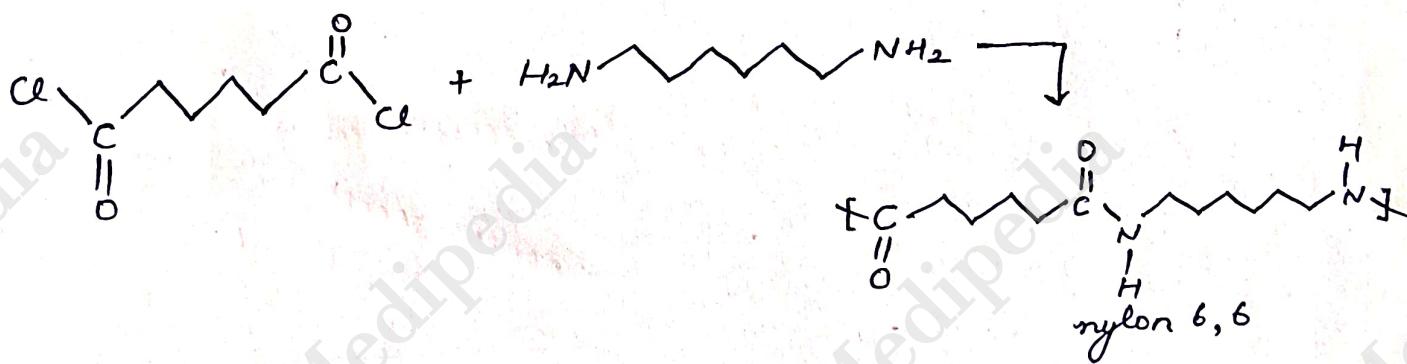
#### a) Additional polymers

They are formed by the repeated addition of monomer molecules forming double or triple bonds.



#### b) Condensation polymers

Condensation polymers are formed by repeated condensation reaction between two different bi-functional or tri-functional monomeric units.



### 4. Based on Molecular force

#### a) Nylon

These are the condensation polymers of diamine and dibasic acids. They have a protein-like structure.

ex. Nylon 6,6

### b) Thermoplastic polymers:

These are linear or slightly branched long chain polymers, which can be softened on heating and reversibly hardened on cooling repeatedly.

ex. Polyvinyl chloride

## PROPERTIES

- It should be versatile and possess a wide range of mechanical, physical, chemical properties.
- It should be inexpensive and easy to fabricate.
- It should be inert to host tissue and compatible with environment.
- It should be non-toxic and have good mechanical strength and should be easily administered.

## ADVANTAGES

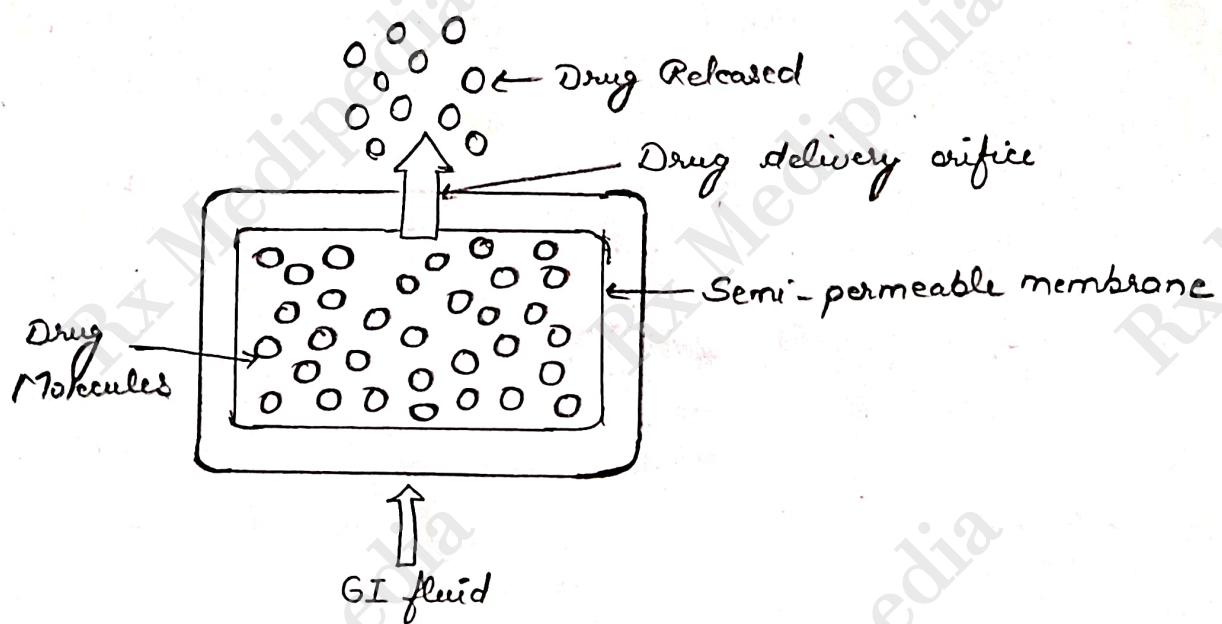
- The polymer can protect the drug from physiological environment and improve its stability.
- The product can be inserted directly at the site where drug action is needed.
- The drug encapsulated is released over long periods.
- Reservoir based polymers increase the solubility of incompletely soluble drugs.

- Gold nanoparticles have capacity to attach with other biomolecules without changing their properties.
- Polymer gave a distinctive property which is not achieved by any of the materials.
- Current polymers [ Poly 2-hydroxy ethyl methacrylate, polyvinyl alcohol] are used because of their inert characteristics.
- Dexton is the common polymer used for coating of Iron oxide.

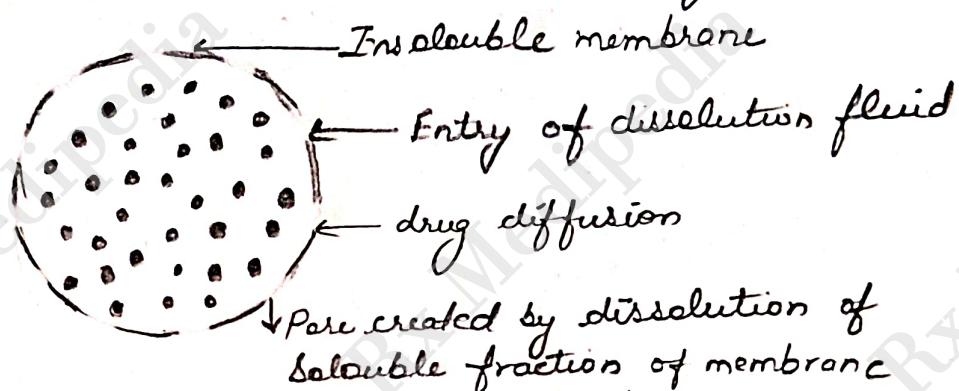
## APPLICATIONS OF POLYMERS

### 1. Osmotic pressure - Controlled GI delivery system

In this device, an osmotic agent is contained within a fixed container. One wall of fixed container is made of semi-permeable membrane. When drug interact with Gastro-intestinal fluids, fluids will force the active agent out of device through the delivery orifice.



2. Gel diffusion controlled Gastro-intestinal delivery System
- In this drug is covered in a partially soluble membrane.
  - Pores are created due to dissolution parts of membrane.
  - It permits entry of aqueous medium in drug dissolution.

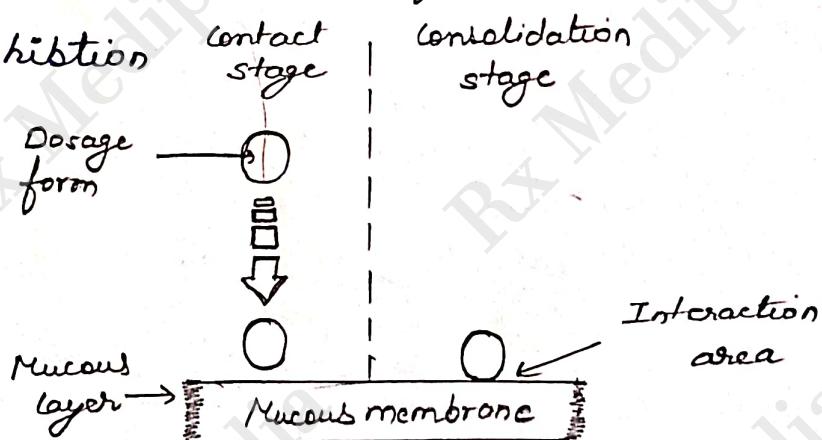


### 3. Mucoadhesive Gastro-intestinal delivery System

Mucoadhesive polymers are used for buccal drug delivery.

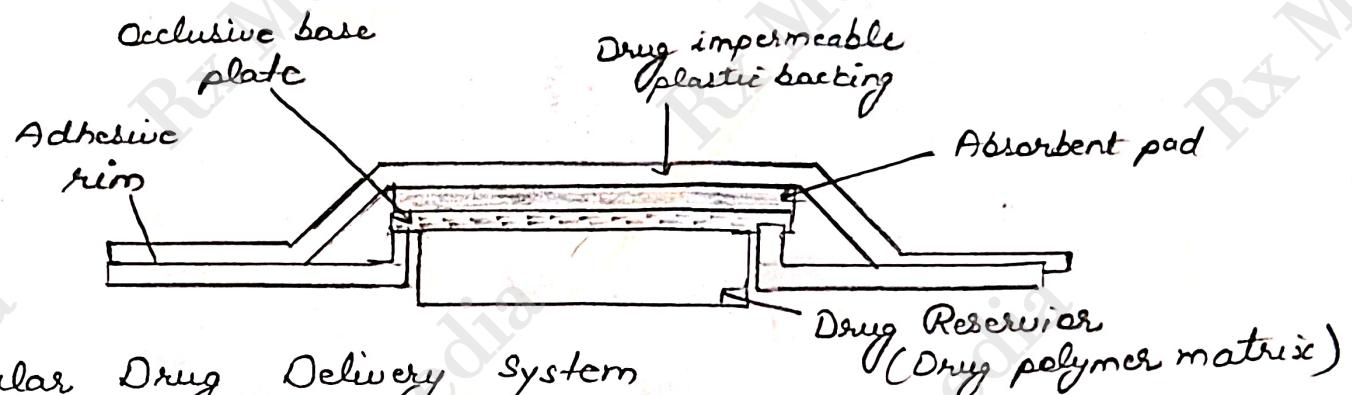
advantages →

- increase in residence time of polymer
- Enzymatic inhibition



### 4. Transdermal Drug delivery System

Transdermal drug delivery System is a self-contained dosage forms, which apply to intact (unbroken) skin, which delivers the drug at a controlled rate to the systemic circulation.



### 5. Ocular Drug Delivery System

It allows prolonged contact of drug with corneal surface of eye. ex. pilocarpine in treatment of glaucoma.

In this, Mucoadhesive polymers are used as a barrier to control the drug release.

ex. Polyacrylic Acid

